



Estd. 1962
NAAC 'A' Grade

SHIVAJI UNIVERSITY, KOLHAPUR-416 004. MAHARASHTRA

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दुरध्वनी (ईपीएबीएक्स) २६०९०००० (अभ्यास मंडळे विभाग— २६०९०९४)

फॅक्स : ००९१-०२३१-२६९१५३३ व २६९२३३३. e-mail: bos@unishivaji.ac.in

SU/BOS/Science & Technology /

Date : 13 JAN 2020

No 0382 →

The Principal,
All Affiliated Colleges/Institutions
Shivaji University, Kolhapur.

Subject: Regarding minor changes in the Chemistry of B.Sc. Part-II Sem- IV, Paper VII & Practical under the Faculty of Science & Technology.

Sir/Madam,

With reference to the subject mentioned above, I am directed to inform you that the Hon'ble Vice-Chancellor have accepted and granted approval to the minor changes in the (Chemistry) of B.Sc.Part-II Sem-IV Paper VII & Practical as per enclosed under the Faculty of Science and Technology.

This minor changes will be implemented from the second term of academic year 2019-2020 onwards.

You are therefore, requested to bring this to the notice of all students and teachers concerned.

Thanking you,

Yours faithfully,

Dy. Registrar

Copy to :-

1	The Dean, Faculty of Science & Technology	8	Appointment Section
2	The Chairman, Respective BOS	9	Centre for Distance Education
3	Exam Section	10	Computer Centre
4	Eligibility Section	11	Affiliation Section (U.G.)
5	B.Sc/M.Sc Section	12	Affiliation Section (P.G.)
6	O.E. II Section	13	P.G.Admission Section
7	O.E. III Section	14	P.G.Seminar Section



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B.Sc. II (CBCS) Sem IV Paper VII DSC-D3 (Inorganic Chemistry)		
	Incorrect	Correct
Title of Paper	Industrial Chemistry	Inorganic Chemistry
Unit No. I. 1.1	Introduction-Defination and formation of co-ordinate covalent in $\text{BF}_3\text{-NH}_3\text{e}[\text{NH}_4]^+$ and H_2O	Introduction-Defination and formation of co-ordinate covalent in $\text{BF}_3\text{-NH}_3$ and $[\text{NH}_4]^+$
Unit No. III. 3.2	Characteristics of p block elements with special reference to Electronic configuration and Periodic properties. reactivity.	Characteristics of p block elements with special reference to Electronic configuration, Atomic radii, Ionic, oxidation state, Ionisation potential, Electronegativity and reactivity.
3.3	Group 13, 14 and 15	Group 13, 14 and 15 only.
Practical 3) Titrimetric Analysis	(Calibration of burette, pipette and volumetric flask is essential)	(Calibration of burette, pipette and volumetric flask must be done by faculty members, General procedure of calibration of burette, pipette & volumetric flasks may be demonstrated to the students.)
v)	Vinegar analysis: To estimate amount of acetic acied from vinegar sample	percentage purity of Ferrous ammonium sulphate